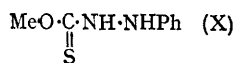
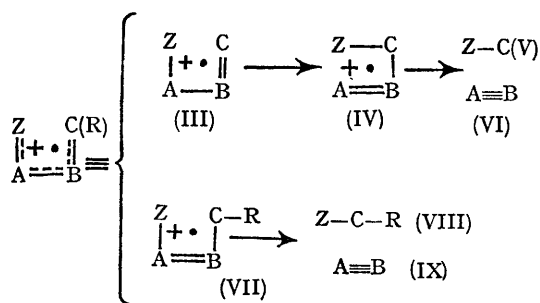
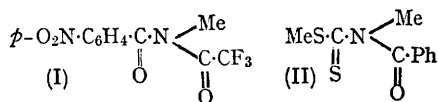


Rationalisation of a Variety of Electron-Impact Induced Rearrangements in Ions

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THE manual, and ultimately automated, interpretation of mass spectra would be simplified by classification of skeletal rearrangement processes. Here, we describe one such class of reactions in conjugated π -systems.



Electron-impact induced aryl migration from nitrogen to oxygen in amides has been described by us,¹ and we have now observed a corresponding aryl migration in the imide (I) to give an ion at m/e 110 ($\text{C}_8\text{H}_3\text{F}_3\text{N}$). Similarly, the loss of carbon dioxide from cyclic imides^{2a,3} may be regarded as occurring after aryl migration, again from nitrogen to oxygen. A corresponding aryl migration can be seen in *N*-phenylphthalimide, which gives a weak ion at m/e 130 ($\text{C}_8\text{H}_4\text{NO}$). In *N*-anilino-phthalimide, a migration of an anilino-group must occur in the formation of an ion at m/e 130 ($\text{C}_8\text{H}_4\text{NO}$). Aryl migration from nitrogen to sulphur has been observed⁴ in the thiocarbamate (II), and migration of a thioaroyl group from nitrogen to sulphur, followed by ejection of CS_2 , in *N*-methylthiophthalimide has been reported.⁵

As a formal generalisation of these and other results, one may consider that rearrangement occurs in the basic π -system (III; A,B,C providing p_π orbitals), in which Z represents any grouping also having a p_π orbital, *e.g.* methylene,⁶ aryl, aroyl,^{2a,3,4} thioaroyl,⁵ anilino, sulphonyl,⁶ and tropylium⁶ groups. Apart from hydrogen, which constitutes a special case,[†] no migration of a group not possessing a p_π orbital has been observed. In the

[†] Rearrangements through cyclic transition states, involving hydrogen transfer, are relatively common. (K. Biemann in "Mass Spectrometry, Organic Chemical Applications", McGraw Hill, New York, 1962, p. 77; F. W. McLafferty in "Mass Spectrometry of Organic Ions", Academic Press, New York, 1963, p. 331).

products [(V), (VI)] of fragmentation of the rearranged ion (IV), ZC is normally ejected as a neutral moiety with AB bearing the positive charge. Ejection of an aryloxy-radical from aromatic azoxy-compounds⁷ may also fit this classification.

Phenyl migration in the closely similar π -system (VII) to give the charged fragment (VIII) and the neutral fragment (IX) can be seen in the mass spectra of benzophenone oxime,^{2b} *O*-methylacetophenone oxime,⁶ and 1,1-dicyano-2-phenylethylene.⁸ In system (III), where A is also bonded to hydrogen, migration both of H and Z to C can be observed, as for example in the thiocarbamate (X) which ejects $C_2H_3NO^4$ and *N*-trifluoroacetylaniline, which yields an ion corresponding to phenol; this

rearrangement has been considered to proceed *via* the tautomer (VII; R = H).⁴

Evidence for the participation of an electronically excited four-centre transition state, at least in some cases, has been presented elsewhere,^{1,3,9} and supplements the five- and six-centre excited-state reactions already described by us.¹⁰

The extent to which skeletal rearrangement processes can be observed in the systems [(III), (VII)] is partly governed by the ease of competing simple cleavage reactions so that in some cases no rearrangement is found.^{6,11} However, where the basic four-centre π -system [(III), (VII)] exists in a molecule, rearrangement of the molecular ion should be sought.

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⁹ R. A. W. Johnstone and D. W. Payling, *Chem. Comm.*, 1968, 601.

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¹¹ C. Nolde, S.-O. Lawesson, J. H. Bowie, and R. G. Cooks, *Tetrahedron*, 1968, **24**, 1051.